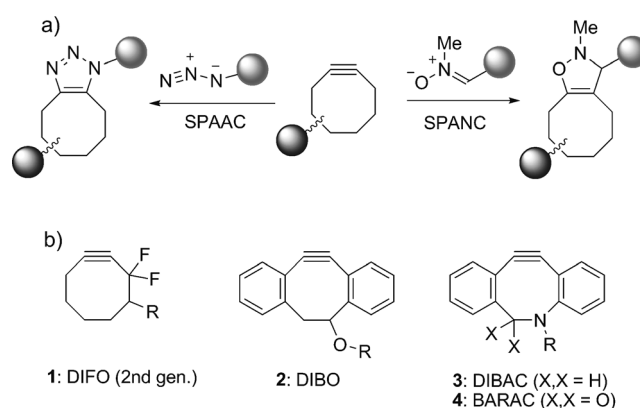


Readily Accessible Bicyclononynes for Bioorthogonal Labeling and Three-Dimensional Imaging of Living Cells**

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The advent of chemical biology tools for imaging and tracking of biomolecules (proteins, lipids, glycans) in their native environment is providing unique insights into cellular processes that are not achievable with traditional biochemical or molecular biology tools.^[1] Bioorthogonal labeling of biomolecules has proven particularly useful for the detection and study of glycans^[2] and lipids,^[3] based on a highly selective reaction between an abiotic functional tag and a designed chemical probe. With respect to the abiotic tag, azide has been used extensively because of its straightforward chemical introduction, small size, and relative inertness.^[4] The finding that azides react rapidly and cleanly with terminal acetylenes in the presence of copper(I), the quintessential “click” reaction, has found tremendous application in life and material sciences.^[5] However, because up to 20 mol % of copper(I) species is typically used, such click chemistry is not suitable for labeling of living systems without compromising cell function.^[6] Apart from that, the presence of copper may induce oligonucleotide^[7] and polysaccharide^[8] degradation. To avoid the use of toxic metals, several metal-free bioorthogonal labeling reaction have been developed.^[9] In particular, phosphines have been used for covalent ligation to azides, a procedure known as Staudinger ligation.^[10] However, owing to the oxygen sensitivity of phosphines, recent focus of chemical ligation is shifting towards strain-promoted



Scheme 1. Reactions and structures of cyclooctyne compounds for strain-promoted cycloaddition. a) Cycloaddition with azide (SPAAC) or nitron (SPANC). b) Structures of the most commonly employed cyclooctynes.

cycloaddition reactions with cyclooctynes (Scheme 1a).^[11] Most prominently, azides were found to react with cyclooctynes with high reaction rates in a so-called strain-promoted alkyne-azide cycloaddition (SPAAC).^[12] The toolbox of metal-free bioorthogonal reactions was most recently further expanded by our research group^[13] and others,^[14] by demonstrating that cyclooctynes undergo even more rapid strain-promoted cycloaddition with nitrones (SPANC), a procedure that was found suitable for dual, irreversible, and site-specific N-terminal modification of proteins.^[13]

The broad application of metal-free cycloaddition in life sciences is, however, hampered by the limited commercial availability and lengthy synthetic routes for preparation of the most common cyclooctynes (Scheme 1b). For example, eight synthetic steps are required to generate second-generation DIFO (**1**),^[15] nine steps for DIBAC (**3**),^[16] and seven steps for BARAC (**4**),^[17] while yields are usually low (10% for **2**,^[18] 16% for **4**). Additional modifications, such as dibenzoannulation (compounds **2–4**), increase lipophilicity and may, therefore lead to non-specific binding to proteins.^[17]

Here we report bicyclo[6.1.0]nonyne (BCN) as a novel ring-strained alkyne for metal-free cycloaddition reactions with azides and nitrones. Bicyclononyne derivatives, which were obtained in a highly straightforward process through cyclopropanation of 1,5-cyclooctadiene, are C_s symmetrical and display excellent reaction kinetics in strain-promoted cycloaddition reactions. Functionalized derivatives of BCN were applied in the labeling of proteins and glycans, as well as in the three-dimensional visualization of living melanoma cells.

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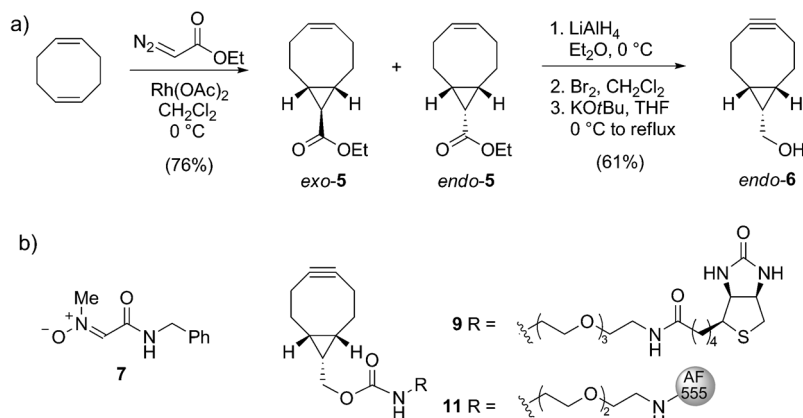
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Based on the known reactivity enhancement of cyclopropane fusion,^[19] we speculated that analogues of bicyclo[6.1.0]nonyne (compound **6**, Scheme 2a) would form a class

The usefulness of BCN for bioorthogonal functionalization of biomolecules was next investigated in the one-pot SPANC functionalization of a model peptide with an N-terminal serine. Indeed, we observed a clean conversion of FRATide, a GSK-1 binding peptide that prevents axin binding,^[22] into the expected isoxazoline conjugates upon SPANC labeling with *endo*-**6** or biotinylated BCN-conjugate **9**, as judged by mass spectrometric analysis and spot-blot analysis (see the Supporting Information).



Scheme 2. a) Synthesis of 9-hydroxymethylbicyclo[6.1.0]nonyne (*endo*-**6**). b) Structures of nitron **7** and BCN conjugated to biotin (**9**) or Alexa Fluor 555 (**11**). THF = tetrahydrofuran.

of versatile cycloalkynes for bioconjugation by combining relative stability with high reactivity. Thus, the synthesis of BCN started by the dropwise addition of ethyl diazoacetate to a large excess (8 equiv) of 1,5-cyclooctadiene in the presence of rhodium acetate.^[20] The resulting mixture of diastereomeric compounds *exo*-**5** and *endo*-**5**, formed in a 2:1 ratio, was readily separated by chromatography on silica gel (combined yield 76%). Next, as exemplified for *endo*-**5**, the individual stereoisomers were converted into the corresponding hydroxyalkynes by reduction of the ester group, bromination, and elimination—a three-step reaction sequence that was performed within eight hours and required only a single purification step. The desired bicyclo[6.1.0]non-4-yn-9-ol (*endo*-**6**) was thus obtained in 61% overall yield after purification (the diastereomeric *exo*-isomer of **6** was prepared in 53% yield from *exo*-**5**). Both *exo*-**6** or *endo*-**6** were found sufficiently stable for prolonged storage at -20°C and did not undergo structural change upon stirring in the presence of 5 mM glutathione for 48 hours in $\text{CD}_3\text{CN}/\text{D}_2\text{O}$ (1:2).^[17]

Next, we investigated the reaction kinetics of **6** in the prototypical SPAAC reaction with benzyl azide. Calculation of second-order reaction kinetics demonstrated that the three-membered ring fusion leads to a near 100-fold rate enhancement over plain cyclooctyne ($k \approx 2 \times 10^{-3} \text{ M}^{-1} \text{ s}^{-1}$), measuring $0.14 \text{ M}^{-1} \text{ s}^{-1}$ for *endo*-**6** and $0.11 \text{ M}^{-1} \text{ s}^{-1}$ for *exo*-**6**. Reaction rates increased, as anticipated,^[13] in a more polar mixture ($\text{CD}_3\text{CN}/\text{D}_2\text{O}$ (1:2)), measuring 0.29 and $0.19 \text{ M}^{-1} \text{ s}^{-1}$ for *endo*-**6** and *exo*-**6**, respectively, values similar to or better than other cyclooctyne systems.^[4,21] Strain-promoted acetylene-nitron cycloaddition (SPANC)^[13] with nitron **7** (Scheme 2b) instead of an azide was found to be significantly faster, measuring 1.66 and $1.32 \text{ M}^{-1} \text{ s}^{-1}$ for *endo*-**6** and *exo*-**6**, respectively.

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The bioavailability and tolerability of labeling surface glycans on living human melanoma MV3 cells was addressed using the chemical reporter strategy.^[1b] MV3 melanoma cells

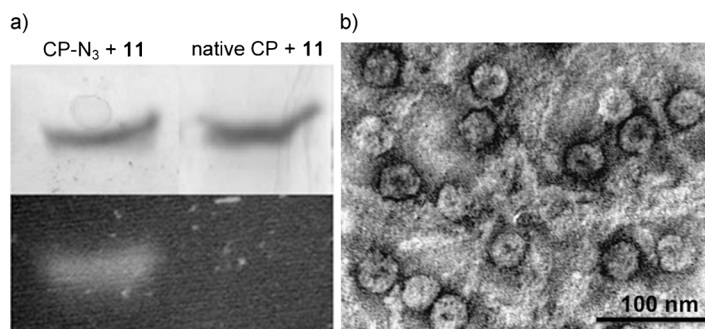


Figure 1. BCN modification of azido-containing virus capsid protein and assembly into virus capsids. a) SDS-PAGE analysis of reaction of BCN-AF555 conjugate (**11**) with capsid protein containing azide (left) or without azide (right). Top: Coomassie Brilliant Blue staining, bottom: fluorescence image. b) TEM pictures of fluorescent capsids (images recorded on a JEOL 1010 TEM, the sample was deposited on a hydrophilized Formvar carbon-coated TEM grid and consequently negatively stained with 0.2% uranyl acetate).

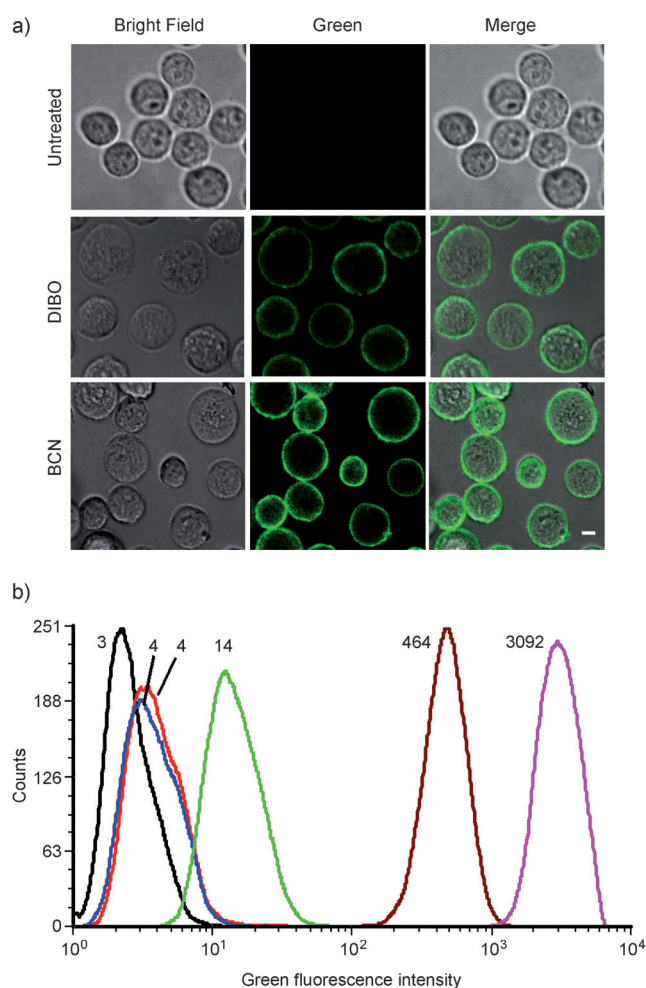


Figure 2. Surface and total fluorescence intensity of MV3 melanoma, cultured in the absence or presence of Ac_4ManNAz ($50\ \mu\text{M}$), followed by labeling with a cyclooctyne-biotin conjugate and secondary labeling with streptavidin-Alexa Fluor 488. a) Representative confocal images of unlabeled cells (top), cells labeled with DIBO-biotin (middle) or BCN-biotin **9** (bottom). Bar: $10\ \mu\text{m}$. b) Label intensity assessed by flow cytometry, indicated as mean fluorescence intensity (MFI). Numbers denote the average of green fluorescent cells for that particular experiment. Black trace: untreated; red trace: Ac_4ManNAz + SA-AF488; blue trace: w/o Ac_4ManNAz + DIBO + SA-AF488; dark red trace: Ac_4ManNAz + DIBO + SA-AF488; green trace: w/o Ac_4ManNAz + BCN + SA-AF488; magenta trace: Ac_4ManNAz + BCN + SA-AF488.

are highly invasive and metastatic, and their abundant production of surface glycans was previously implicated in invasion processes.^[24] Thus, MV3 cells were incubated with peracetylated *N*-azidoacetyl-D-mannosamine (Ac_4ManNAz), labeled with BCN-biotin conjugate **9**, and stained with streptavidin-Alexa Fluor 488 (Figure 2). To be able to compare the efficiency of labeling of **9** to that of dibenzocyclooct-

tyne (DIBO, **2**), one of the most reactive cyclooctyne systems known to date,^[18] cells were also labeled with a DIBO-biotin conjugate. In all cases, cells retained morphological integrity and cell surface fluorescence, with consistently higher labeling for BCN than for DIBO, as detected by confocal microscopy (Figure 2a) and flow cytometry (Figure 2b). By using flow cytometry, high monophasic intensities and excellent signal-to-noise ratio (SNR) were found for both DIBO-biotin (SNR = 116) and BCN-biotin (SNR = 221). No signs of label-induced cytotoxicity were detected after propidium iodide staining (see the Supporting Information).

Functional cell integrity was confirmed after incorporating MV3 cells into three-dimensional collagen lattices yielding spontaneous and vigorous invasion. Owing to its high signal-to-noise ratio, labeling with BCN revealed fine, sub-cellular details, which can discriminate surface glycan distribution states on individual living cells as shown by densitometry experiments (see the Supporting Information). Whereas cells in suspension retain a near-homogeneous distribution of azidosialic acids on the cell surface (as in Figure 2), invading cells show the redistribution and accumulation of sialic acid at actin-rich contact sites with collagen fibers, consistent with their role in cell adhesion and migration (Figure 3).^[25,26,27] Thereby, submicron resolution reveals fine surface distribution of sialic acids at leading edge filopodia, focal clusters at actin-rich contact sites to collagen fibers, and substantial glycan-rich deposits into the tissue matrix from the trailing edge (see the Supporting Information).

In conclusion, in view of the non-toxic labeling procedure, the tunable fluorescent properties by choice of dye and the high signal-to-noise ratio, BCN will be useful for addressing molecular glycan function studies in live-cell and other systems. A key advantage of BCN over earlier cyclooctynes lies in the combination of its exceptionally easy preparation with high reactivity. Furthermore, it should be noted that the lack of conformational isomerism in bicyclo[6.1.0]non-4-ynes^[29] leads to sharp peaks in the ^1H NMR spectrum, which is further simplified by the C_s symmetry of BCN. An additional advantage of a symmetrical cyclooctyne is the

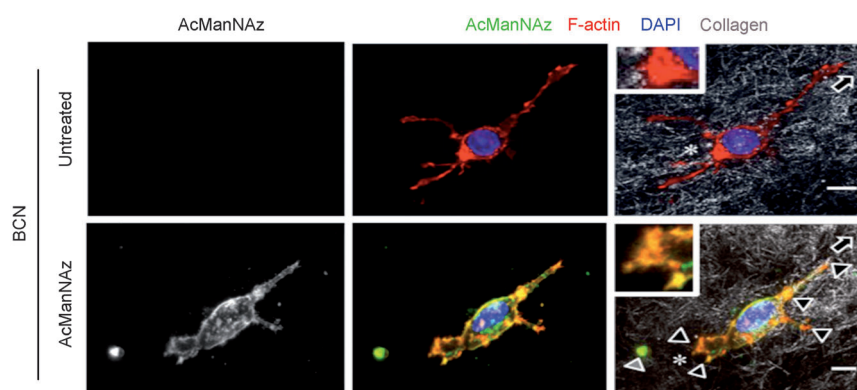


Figure 3. Live-cell staining and redistribution of glycans during invasive cell migration through a three-dimensional collagen matrix. Focal accumulation of sialic acid on migrating MV3 melanoma cell at interaction sites to collagen fibers and partial colocalization with F-actin. Insets, trailing edge. Direction of migration was determined from retraction fibers (asterisks) and deposited sialic acid-rich material lacking F-actin from the cell rear (gray arrowhead).^[28] Bar: $5\ \mu\text{m}$. Focalized glycan distribution at cell matrix interactions (black arrowheads).

formation of a single regioisomer upon cycloaddition, an aspect of particular advantage in areas where the formation of homogeneous adducts is mandatory.^[11] Thus, BCN will allow a broad range of applications that require the highly efficient and metal-free conjugation of two separate molecular entities, for example in life sciences, material science, surface modification, and molecular diagnostics.

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- [1] a) M. D. Best, *Biochemistry* **2009**, *48*, 6571–6584; b) E. M. Sletten, C. R. Bertozzi, *Angew. Chem.* **2009**, *121*, 7108–7133; *Angew. Chem. Int. Ed.* **2009**, *48*, 6974–6998.
- [2] J. A. Prescher, C. R. Bertozzi, *Nat. Chem. Biol.* **2005**, *1*, 13–21.
- [3] G. Charron, J. Wiljon, H. C. Hang, *Curr. Opin. Chem. Biol.* **2009**, *13*, 382–391.
- [4] M. F. Debets, C. W. J. van der Doelen, F. P. J. T. Rutjes, F. L. van Delft, *ChemBioChem* **2010**, *11*, 1168–1184.
- [5] M. Meldal, C. W. Tornøe, *Chem. Rev.* **2008**, *108*, 2952–3015.
- [6] A. J. Link, M. K. S. Vink, D. A. Tirrell, *J. Am. Chem. Soc.* **2004**, *126*, 10598–10602.
- [7] J. Gierlich, G. A. Burley, P. M. E. Gramlich, D. M. Hammond, T. Carell, *Org. Lett.* **2006**, *8*, 3639–3642.
- [8] E. Lallana, E. Fernandez-Megia, R. Rigueru, *J. Am. Chem. Soc.* **2009**, *131*, 5748–5750.
- [9] C. R. Becer, R. Hooogenboom, U. Schubert, *Angew. Chem.* **2009**, *121*, 4998–5006; *Angew. Chem. Int. Ed.* **2009**, *48*, 4900–4908.
- [10] M. Köhn, R. Breinbauer, *Angew. Chem.* **2004**, *116*, 3168–3178; *Angew. Chem. Int. Ed.* **2004**, *43*, 3106–3116.
- [11] J.-F. Lutz, *Angew. Chem.* **2008**, *120*, 2212–2214; *Angew. Chem. Int. Ed.* **2008**, *47*, 2182–2184.
- [12] a) A. T. Blomquist, L. H. Liu, *J. Am. Chem. Soc.* **1953**, *75*, 2153–2154; b) N. J. Agard, J. A. Prescher, C. R. Bertozzi, *J. Am. Chem. Soc.* **2004**, *126*, 15046–15047.
- [13] X. Ning, R. P. Temming, J. Dommerholt, J. Guo, D. B. Ania, M. F. Debets, M. A. Wolfert, G. J. Boons, F. L. van Delft, *Angew. Chem.* **2010**, *122*, 3129–3132; *Angew. Chem. Int. Ed.* **2010**, *49*, 3065–3068.
- [14] C. S. McKay, J. Moran, J. P. Pezacki, *Chem. Commun.* **2010**, *46*, 931–933.
- [15] J. A. Codelli, J. M. Baskin, N. J. Agard, C. R. Bertozzi, *J. Am. Chem. Soc.* **2008**, *130*, 11486–11493.
- [16] M. F. Debets, S. S. van Berkel, S. Schoffelen, F. P. J. T. Rutjes, J. C. M. van Hest, F. L. van Delft, *Chem. Commun.* **2010**, *46*, 97–99.
- [17] J. C. Jewett, E. M. Sletten, C. R. Bertozzi, *J. Am. Chem. Soc.* **2010**, *132*, 3688–3690.
- [18] X. Ning, J. Guo, M. A. Wolfert, G.-J. Boons, *Angew. Chem.* **2008**, *120*, 2285–2287; *Angew. Chem. Int. Ed.* **2008**, *47*, 2253–2255.
- [19] H. Meier, C. Schuh-Popitz, H. Peiersen, *Angew. Chem.* **1981**, *93*, 286–287; *Angew. Chem. Int. Ed. Engl.* **1981**, *20*, 270–271.
- [20] *Modern Rhodium-Catalyzed Organic Reactions* (Ed.: P. A. Evans), Wiley-VCH, Weinheim, **2005**.
- [21] N. J. Agard, J. M. Baskin, J. A. Prescher, A. Lo, C. R. Bertozzi, *ACS Chem. Biol.* **2006**, *1*, 644–648.
- [22] G. M. Thomas, S. Frame, M. Goedert, I. Nathke, P. Polakis, P. Cohen, *FEBS Lett.* **1999**, *458*, 247–251.
- [23] A. J. Link, D. A. Tirrell, *Methods* **2005**, *36*, 291–298.
- [24] M. Goebeler, D. Kaufmann, E. B. Brocker, C. E. Klein, *J. Cell Sci.* **1996**, *109*, 1957–1964.
- [25] C. H. Chiang, C. H. Wang, H. C. Chang, S. V. More, W. S. Li, W. C. Hung, *J. Cell. Physiol.* **2010**, *223*, 492–499.
- [26] D. R. Christie, F. M. Shaikh, J. A. Lucas IV, J. A. Lucas III, S. L. Bellis, *J. Ovarian Res.* **2008**, *1*, 3.
- [27] E. C. Seales, G. A. Jurado, B. A. Brunson, J. K. Wakefield, A. R. Frost, S. L. Beilis, *Cancer Res.* **2005**, *65*, 4645–4652.
- [28] C. Mayer, K. Maaser, N. Daryab, K. S. Zanker, E. B. Brocker, P. Friedl, *Eur. J. Cell Biol.* **2004**, *83*, 709–715.
- [29] C. Antony-Mayer, H. Meier, *Chem. Ber.* **1988**, *121*, 2013–2018.